

Aquatic Toxicity of Common Household Chemicals and Stormwater

Team: Amanda Seidler & Lydia White
 Mentor: Dr. Felicia Armstrong
 Department of Geological & Environmental Sciences



Abstract

Toxicity testing is a process by which environmental conditions can be evaluated. *Daphnia magna* is a standard test organism to test aquatic environments. *Daphnia* are used because they are easily cultured in the lab and sensitive to a variety of contaminants and pollutants. Common contaminants such as rock salt and fertilizers were used in 24 hour acute toxicity tests. These chemicals are commonly used in urban and suburban environments without much thought of their environmental effects. *Daphnia* were acclimated to moderately hard water following USEPA guidelines prior to testing. Using the same water, chemicals were added at multiple concentrations and replicated three times at each level including an uncontaminated control. *Daphnia* were exposed for 24 hours while being monitored. Counts were done every couple of hours for live *Daphnia* and dead *Daphnia* were removed. Water test were done for dissolved oxygen, pH, conductivity, hardness and temperature to insure proper conditions. At the end of the test period, the percent mortality was determined and plotted against concentration.

Many of the chemicals and waste that humans produce can adversely effect wildlife. One of the most impacted systems are aquatic freshwater systems. Water from land runoff, sewage discharge, industrial waste, atmospheric deposition, and many other sources can contaminate water and make it unsuitable for aquatic life. Many chemicals that are used on lawns, gardens or in agriculture have been determined to be toxic to aquatic life therefore application is not recommend near surface waters or prior to rainfall which could move the chemicals into the streams. Understanding the toxicity of various common chemicals and storm water will enable better decisions made on the use or avoidance of these chemicals in our lives.

Methodology

Dilution Water

- ‡ Water was used in tank and as the dilution water for the testing solutions
- ‡ pH: 7.4-7.8, hardness: 8000, alkalinity: 5764

Collection of Storm Water

- ‡ Water was collected at a point of interest during rain events.
- ‡ The rain event was at least 72 hours after the last event
- ‡ One liter samples were collected in clean plastic containers
- ‡ Samples were collected at storm drains

Preparation of Testing Water

- ‡ Dilutions (250 mL) were made at varying concentrations based on literature findings
- ‡ Approximately 50 mL of testing solution were added to each container

Preparation of Daphnia for Testing

- ‡ *Daphnia* they were acclimated to the synthetic freshwater for a minimum of 48 hours
- ‡ Microscope selection of *Daphnia* was used to identify full sized, non-pregnant *Daphnia*
- ‡ Ten *Daphnia* were transferred by pipet, to each concentration testing container

Testing

- ‡ *Daphnia* were subjected to 24 hour static toxicity test
- ‡ *Daphnia* were monitored every couple of hours for mortality
- ‡ Dead *Daphnia* were removed from the containers

Results & Discussion

Table 2 Deicer (Brine) Water Chemical Analysis

| Water | pH | Conductivity (uS/cm) | DO (mg/L) |
|-------------|------|----------------------|-----------|
| Deicer 0.3% | 7.10 | 1907 | 7.2 |
| Deicer 0.5% | 7.02 | 2910 | 7.2 |
| Deicer 1.0% | 7.00 | 5290 | 7.2 |

Table 4 Storm Water Chemical Analysis

| Water | pH | Conductivity (uS/cm) | DO (mg/L) |
|------------------|------|----------------------|-----------|
| Storm Water 25% | 6.96 | 478 | 6.7 |
| Storm Water 50% | 7.44 | 611 | 6.7 |
| Storm Water 100% | 7.91 | 883 | 6.7 |

Table 3 Bayer Water Chemical Analysis

| Water | pH | Conductivity (uS/cm) | DO (mg/L) |
|--------------|------|----------------------|-----------|
| Bayer 0.025% | 6.75 | 444 | 8.6 |
| Bayer 0.050% | 6.83 | 558 | 8.6 |
| Bayer 0.100% | 6.97 | 767 | 8.6 |

Table 5 Dilution Water Chemical Analysis

| Water | pH | Conductivity (uS/cm) | DO (mg/L) |
|----------------|-----|----------------------|-----------|
| Dilution Water | 8.0 | 327 | 8.75 |

- ‡ Safer is a insecticidal soap that uses potassium salts of fatty acid as the active ingredient
- ‡ All concentrations of the insecticidal soap showed signs of toxicity to *Daphnia*
- ‡ Insecticidal soaps are targeted to kill pest such as Aphids which come from the same phylum (Arthropoda) as *Daphnia*
- ‡ Bayer is weed killer used to kill grass, dandelions and clovers. It contains 2,4-D, dimethylamine salt (4.85%), Quinclorac (1.61%) and Dicamba dimethylamine salt (0.45%) as the active ingredients.
- ‡ Bayer pesticide had the highest *Daphnia* mortality for this toxicity test.
- ‡ Aqua Salina, a deicer/anti-ice is brine water formed from conventional wells.
- ‡ *Daphnia* were relatively resistant to the Deicer (brine) water with mortality not occurring until end of the 24 hour toxicity test.
- ‡ Hubbard Ohio storm water was collected from a drain carrying suburban runoff. It did not show high amounts of toxicity.
- ‡ According to the calculated LC50 the Safer insecticide was the most toxic followed very closely by Bayer herbicide followed by the Deicer.
- ‡ All these solutions caused mortality at less than 1%.

Conclusion

Acute toxicity testing of *Daphnia magna* led to overall decline in health and resulted in mortality. There was higher decline of health and mortality in the low concentrations of both the pesticide and herbicide. Safer pesticide solution had the greatest toxicity at the lowest concentration. Even at the lowest concentration, these products can still lead to adverse effect on aquatic organisms. These are only some of the household chemicals that enter aquatic systems through runoff and can be detrimental to aquatic life. Such products should not be used around aquatic ecosystems.

References

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